

## Determination of the amount of lead in water using double isotope dilution and inductively coupled plasma mass spectrometry

This is the example A7 of the EURACHEM / CITAC Guide "Quantifying Uncertainty in Analytical Measurement", Second Edition.

The amount content of lead in water is measured using Isotope Dilution Mass Spectrometry (IDMS) and Inductively Coupled Plasma Mass Spectrometry (ICP-MS)

In this case a 'double' isotope dilution is applied. It uses a well characterised (ideally certified) material of natural isotopic composition as a primary assay standard. Two blends are then prepared: blend b, which is a blend between known masses of the sample and the enriched spike, and blend b', which is the blend between the enriched spike and the primary assay standard. The isotope ratios of the primary assay standard, the spike, the sample and the two blends are measured using ICP-MS. Together with the weighing data of the blends, the amount content of lead in the sample can be calculated.

### Model Equation:

{equation for the double isotope dilution}

$$c_x = (c_z * m_{y1} / m_x * m_z / m_{y2} * (K_{y1} * R_{y1} - K_{b1} * R_{b1}) / (K_{b1} * R_{b1} - K_{x1} * R_{x1}) * (K_{b2} * R_{b2} - K_{z1} * R_{z1}) / (K_{y1} * R_{y1} - K_{b2} * R_{b2}) / (\sum K_{zi} R_{zi}) * (\sum K_{xi} R_{xi})) - c_{blank};$$

$$\sum K_{xi} R_{xi} = K_{x1} * R_{x1} + K_{x2} * R_{x2} + K_{x3} * R_{x3} + K_{x4} * R_{x4};$$

$$\sum K_{zi} R_{zi} = K_{z1} * R_{z1} + K_{z2} * R_{z2} + K_{z3} * R_{z3} + K_{z4} * R_{z4};$$

{calculation of the molar mass of the lead of the primary assay standard 1}

$$M_{Pb \text{ Assay1}} = (K_{z1} * R_{z1} * M_{z1} + K_{z2} * R_{z2} * M_{z2} + K_{z3} * R_{z3} * M_{z3} + K_{z4} * R_{z4} * M_{z4}) / (\sum K_{zi} R_{zi});$$

{concentration of the primary assay standard z which is used for the double IDMS}

$$c_z = m_2 / d_2 * m_1 * w / d_1 / M_{Pb \text{ Assay1}} * k_{mol};$$

{calculation of the K-factors for the various isotope ratios measured}

$$K_{b1} = K_{0\_b1} + K_{bias\_b1};$$

$$K_{b2} = K_{0\_b2} + K_{bias\_b2};$$

$$K_{x1} = K_{0\_x1} + K_{bias\_x1};$$

$$K_{x2} = K_{0\_x2} + K_{bias\_x2};$$

$$K_{x3} = K_{0\_x3} + K_{bias\_x3};$$

$$K_{x4} = K_{0\_x4} + K_{bias\_x4};$$

$$K_{y1} = K_{0\_y1} + K_{bias\_y1};$$

$$K_{z1} = K_{0\_z1} + K_{bias\_z1};$$

$$K_{z2} = K_{0\_z2} + K_{bias\_z2};$$

$$K_{z3} = K_{0\_z3} + K_{bias\_z3};$$

$$K_{z4} = K_{0\_z4} + K_{bias\_z4};$$

**List of Quantities:**

Quantity	Unit	Definition
$c_x$	$\mu\text{mol/g}$	amount content of the sample x
$c_z$	$\mu\text{mol/g}$	amount content of the primary assay standard z
$m_{y1}$	g	mass of enriched spike in blend b
$m_x$	g	mass of sample in blend b
$m_z$	g	mass of primary assay standard in blend b'
$m_{y2}$	g	mass of enriched spike in blend b'
$K_{y1}$		mass bias correction of $R_{y1}$
$R_{y1}$		measured ratio of enriched isotope to reference isotope in the enriched spike, $n(^{208}\text{Pb})/n(^{206}\text{Pb})$
$K_{b1}$		mass bias correction of $R_{b1}$
$R_{b1}$		measured ratio of blend b, $n(^{208}\text{Pb})/n(^{206}\text{Pb})$
$K_{x1}$		mass bias correction of $R_{x1}$
$R_{x1}$		measured ratio of enriched isotope to reference isotope in the sample x, $n(^{208}\text{Pb})/n(^{206}\text{Pb})$
$K_{b2}$		mass bias correction of $R_{b2}$
$R_{b2}$		measured ratio of blend b', $n(^{208}\text{Pb})/n(^{206}\text{Pb})$
$K_{z1}$		mass bias correction of $R_{z1}$
$R_{z1}$		measured ratio of enriched isotope to reference isotope in the primary assay standard z, $n(^{208}\text{Pb})/n(^{206}\text{Pb})$
$\sum K_{zi}R_{zi}$		sum of all mass bias corrected ratios of the primary assay standard
$\sum K_{xi}R_{xi}$		sum of all mass bias corrected ratios of the sample
$c_{\text{blank}}$	$\mu\text{mol/g}$	observed amount content in procedure blank
$K_{x2}$		mass bias correction of $R_{x2}$
$R_{x2}$		measured ratio of sample, $n(^{206}\text{Pb})/n(^{206}\text{Pb})$
$K_{x3}$		mass bias correction of $R_{x3}$
$R_{x3}$		measured ratio of sample, $n(^{207}\text{Pb})/n(^{206}\text{Pb})$
$K_{x4}$		mass bias correction of $R_{x4}$
$R_{x4}$		measured ratio of sample, $n(^{204}\text{Pb})/n(^{206}\text{Pb})$
$K_{z2}$		mass bias correction of $R_{z2}$
$R_{z2}$		measured ratio of sample, $n(^{206}\text{Pb})/n(^{206}\text{Pb})$
$K_{z3}$		mass bias correction of $R_{z3}$
$R_{z3}$		measured ratio of sample, $n(^{207}\text{Pb})/n(^{206}\text{Pb})$
$K_{z4}$		mass bias correction of $R_{z4}$
$R_{z4}$		measured ratio of sample, $n(^{204}\text{Pb})/n(^{206}\text{Pb})$
$M_{\text{Pb Assay1}}$	g/mol	molar mass of the primary assay standard
$M_{z1}$	g/mol	nuclidic mass of $^{208}\text{Pb}$
$M_{z2}$	g/mol	nuclidic mass of $^{206}\text{Pb}$
$M_{z3}$	g/mol	nuclidic mass of $^{207}\text{Pb}$

Quantity	Unit	Definition
$M_{z4}$	g/mol	nuclidic mass of $^{204}\text{Pb}$
$m_2$	g	aliquot of the first dilution of the primary assay standard
$d_2$	g	total mass of the second dilution of the primary assay standard
$m_1$	g	mass of the lead piece for primary assay standard
$w$	g/g	purity of the metallic lead piece, expressed as mass fraction
$d_1$	g	total mass of first dilution of the primary assay standard
$k_{\text{mol}}$	$\mu\text{mol/mol}$	conversion factor $10^6 \mu\text{mol} = 1 \text{ mol}$
$K_{0\_b1}$		mass bias correction of $R_{b1}$ as determined at time 0
$K_{\text{bias\_}b1}$		other contributions to the mass bias of $R_{b1}$
$K_{0\_b2}$		mass bias correction of $R_{b2}$ as determined at time 0
$K_{\text{bias\_}b2}$		other contributions to the mass bias of $R_{b2}$
$K_{0\_x1}$		mass bias correction of $R_{x1}$ as determined at time 0
$K_{\text{bias\_}x1}$		other contributions to the mass bias of $R_{x1}$
$K_{0\_x2}$		mass bias correction of $R_{x2}$ as determined at time 0
$K_{\text{bias\_}x2}$		other contributions to the mass bias of $R_{x2}$
$K_{0\_x3}$		mass bias correction of $R_{x3}$ as determined at time 0
$K_{\text{bias\_}x3}$		other contributions to the mass bias of $R_{x3}$
$K_{0\_x4}$		mass bias correction of $R_{x4}$ as determined at time 0
$K_{\text{bias\_}x4}$		other contributions to the mass bias of $R_{x4}$
$K_{0\_y1}$		mass bias correction of $R_{y1}$ as determined at time 0
$K_{\text{bias\_}y1}$		other contributions to the mass bias of $R_{y1}$
$K_{0\_z1}$		mass bias correction of $R_{z1}$ as determined at time 0
$K_{\text{bias\_}z1}$		other contributions to the mass bias of $R_{z1}$
$K_{0\_z2}$		mass bias correction of $R_{z2}$ as determined at time 0
$K_{\text{bias\_}z2}$		other contributions to the mass bias of $R_{z2}$
$K_{0\_z3}$		mass bias correction of $R_{z3}$ as determined at time 0
$K_{\text{bias\_}z3}$		other contributions to the mass bias of $R_{z3}$
$K_{0\_z4}$		mass bias correction of $R_{z4}$ as determined at time 0
$K_{\text{bias\_}z4}$		other contributions to the mass bias of $R_{z4}$

$m_{y1}$ : Type B normal distribution  
 Value: 1.1360 g  
 Expanded Uncertainty: 0.0002 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**$m_x$ :** Type B normal distribution  
 Value: 1.0440 g  
 Expanded Uncertainty: 0.0002 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**$m_z$ :** Type B normal distribution  
 Value: 1.1029 g  
 Expanded Uncertainty: 0.0002 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**$m_{y2}$ :** Type B normal distribution  
 Value: 1.0654 g  
 Expanded Uncertainty: 0.0002 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**$R_{y1}$ :** Type A summarized  
 Mean: 0.00064  
 Standard Uncertainty:  $=0.00004/\sqrt{8}$   
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by  $\sqrt{8}$ .

**$R_{b1}$ :** Type A summarized  
 Mean: 0.29360  
 Standard Uncertainty:  $=0.00073/\sqrt{8}$   
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by  $\sqrt{8}$ .

**$R_{x1}$ :** Type A summarized  
 Mean: 2.1402  
 Standard Uncertainty:  $=0.0054/\sqrt{8}$   
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by  $\sqrt{8}$ .

**$R_{b2}$ :** Type A summarized  
 Mean: 0.5050  
 Standard Uncertainty:  $=0.0013/\sqrt{8}$   
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by  $\sqrt{8}$ .

**R<sub>z1</sub>:** Type A summarized  
 Mean: 2.1429  
 Standard Uncertainty: =0.0054/sqrt(8)  
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by sqrt(8).

**C<sub>blank</sub>:** Type A summarized  
 Mean: 4.5·10<sup>-7</sup> μmol/g  
 Standard Uncertainty: =4.0e-7/sqrt(4)  
 Degrees of Freedom: 3

The procedure blank was measured using external calibration. The procedure blank was measured four times. The experimental standard deviation is divided by sqrt(4) to obtain the standard uncertainty.

**R<sub>x2</sub>:** Constant  
 Value: 1

This is the ratio of n(<sup>206</sup>Pb)/n(<sup>206</sup>Pb), which is by definition equal to 1.

**R<sub>x3</sub>:** Type A summarized  
 Mean: 0.9142  
 Standard Uncertainty: =0.0032/sqrt(8)  
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by sqrt(8).

**R<sub>x4</sub>:** Type A summarized  
 Mean: 0.05901  
 Standard Uncertainty: =0.00035/sqrt(8)  
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by sqrt(8).

**R<sub>z2</sub>:** Constant  
 Value: 1

This is the ratio of n(<sup>206</sup>Pb)/n(<sup>206</sup>Pb), which is by definition equal to 1.

**R<sub>z3</sub>:** Type A summarized  
 Mean: 0.9147  
 Standard Uncertainty: =0.0032/sqrt(8)  
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by sqrt(8).

**R<sub>z4</sub>:** Type A summarized  
 Mean: 0.05870  
 Standard Uncertainty: =0.00035/sqrt(8)  
 Degrees of Freedom: 7

Each ratio has been measured 8 times. The experimental uncertainty is therefore divided by sqrt(8).

**M<sub>z1</sub>:** Type B normal distribution  
 Value: 207.976636 g/mol  
 Expanded Uncertainty: 0.000003 g/mol  
 Coverage Factor: 1

The nuclidic masses and their respective uncertainties are taken from literature. G. Audi and A. H. Wapstra, Nuclear Physics, A565 (1993).

**M<sub>z2</sub>:** Type B normal distribution  
 Value: 205.974449 g/mol  
 Expanded Uncertainty: 0.000003 g/mol  
 Coverage Factor: 1

The nuclidic masses and their respective uncertainties are taken from literature. G. Audi and A. H. Wapstra, Nuclear Physics, A565 (1993).

**M<sub>z3</sub>:** Type B normal distribution  
 Value: 206.975880 g/mol  
 Expanded Uncertainty: 0.000003 g/mol  
 Coverage Factor: 1

The nuclidic masses and their respective uncertainties are taken from literature. G. Audi and A. H. Wapstra, Nuclear Physics, A565 (1993).

**M<sub>z4</sub>:** Type B normal distribution  
 Value: 203.973028 g/mol  
 Expanded Uncertainty: 0.000003 g/mol  
 Coverage Factor: 1

The nuclidic masses and their respective uncertainties are taken from literature. G. Audi and A. H. Wapstra, Nuclear Physics, A565 (1993).

**m<sub>2</sub>:** Type B normal distribution  
 Value: 1.0292 g  
 Expanded Uncertainty: 0.0002 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**d<sub>2</sub>:** Type B normal distribution  
 Value: 99.931 g  
 Expanded Uncertainty: 0.01 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**m<sub>1</sub>:** Type B normal distribution  
 Value: 0.36544 g  
 Expanded Uncertainty: 0.00005 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**w:** Type B normal distribution  
 Value: 0.99999 g/g  
 Expanded Uncertainty: 0.000005 g/g  
 Coverage Factor: 1

The purity of the metallic lead can be obtained through analysis or a supplier's certificate.

**d<sub>1</sub>:** Type B normal distribution  
 Value: 196.14 g  
 Expanded Uncertainty: 0.03 g  
 Coverage Factor: 1

Weighings are performed by a dedicated mass metrology lab. The procedure applied was a bracketing technique using calibrated weights and a comparator. The bracketing technique was repeated at least six times for every sample mass determination. Buoyancy correction was applied. The uncertainties from the weighing certificates were treated as standard uncertainties, Type B.

**k<sub>mol</sub>:** Constant  
 Value: 1·10<sup>6</sup> μmol/mol

**K<sub>0\_b1</sub>:** Type A summarized  
 Mean: 0.9987  
 Standard Uncertainty: =0.0025/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_b1</sub>:** Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_b2</sub>:** Type A summarized  
 Mean: 0.9983  
 Standard Uncertainty: =0.0025/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_b2</sub>:** Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_x1</sub>**: Type A summarized  
 Mean: 0.9992  
 Standard Uncertainty: =0.0025/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_x1</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_x2</sub>**: Constant  
 Value: 1

This mass bias correction refers to the ratio of n(<sup>206</sup>Pb)/n(<sup>206</sup>Pb), which is by definition equal to 1 and not measured. Therefore no mass bias correction is needed, the factor is equal to 1.

**K<sub>bias\_x2</sub>**: Constant  
 Value: 0

This mass bias correction refers to the ratio of n(<sup>206</sup>Pb)/n(<sup>206</sup>Pb), which is by definition equal to 1 and not measured. Therefore no mass bias correction is needed, this factor is equal to 0.

**K<sub>0\_x3</sub>**: Type A summarized  
 Mean: 1.0004  
 Standard Uncertainty: =0.0035/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.



**K<sub>bias\_x3</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_x4</sub>**: Type A summarized  
 Mean: 1.001  
 Standard Uncertainty: =0.006/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_x4</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_y1</sub>**: Type A summarized  
 Mean: 0.9999  
 Standard Uncertainty: =0.0025/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_y1</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_z1</sub>**: Type A summarized  
 Mean: 0.9989  
 Standard Uncertainty: =0.0025/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_z1</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_z2</sub>**: Constant  
 Value: 1

This mass bias correction refers to the ratio of n(<sup>206</sup>Pb)/n(<sup>206</sup>Pb), which is by definition equal to 1 and not measured. Therefore no mass bias correction is needed, the factor is equal to 1.

**K<sub>bias\_z2</sub>**: Constant  
 Value: 0

This mass bias correction refers to the ratio of n(<sup>206</sup>Pb)/n(<sup>206</sup>Pb), which is by definition equal to 1 and not measured. Therefore no mass bias correction is needed, this factor is equal to 0.

**K<sub>0\_z3</sub>**: Type A summarized  
 Mean: 0.9993  
 Standard Uncertainty: =0.0035/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_z3</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**K<sub>0\_z4</sub>**: Type A summarized  
 Mean: 1.0002  
 Standard Uncertainty: =0.006/sqrt(8)  
 Degrees of Freedom: 7

The K<sub>0</sub>'s are measured using a certified isotopic reference material, and they are calculated according to the following equation:

$$K_0 = R_{\text{certified}}/R_{\text{observed}}$$

When looking at the uncertainty contributions of R<sub>certified</sub> and R<sub>observed</sub>, it is clear that the contribution of R<sub>certified</sub> can be neglected for this example. Henceforth, the uncertainties on the measured ratios, R<sub>observed</sub>, are used for the uncertainties on K<sub>0</sub>.

The original measurement data for the determination of K<sub>0</sub> is not shown in this example.

**K<sub>bias\_z4</sub>**: Type B normal distribution  
 Value: 0  
 Expanded Uncertainty: 0.001  
 Coverage Factor: 1

This bias factor is introduced to account for possible deviations in the value of the mass discrimination factor (these could be variations over time, as well as other sources of bias, such as multiplier dead time correction, matrix effects etc.). The values of these biases are not known in this case, but they are assumed to be around 0, therefore a value of 0 is applied. An uncertainty is associated to this bias, which is estimated from experience. In this case a standard uncertainty of 0.001 is considered to be sufficient to cover all effects.

**Interim Results:**

Quantity	Value	Standard Uncertainty
$c_z$	0.09260484 $\mu\text{mol/g}$	$27.77 \cdot 10^{-6} \mu\text{mol/g}$
$K_{y1}$	0.999900	$1.335 \cdot 10^{-3}$
$K_{b1}$	0.998700	$1.335 \cdot 10^{-3}$
$K_{x1}$	0.999200	$1.335 \cdot 10^{-3}$
$K_{b2}$	0.998300	$1.335 \cdot 10^{-3}$
$K_{z1}$	0.998900	$1.335 \cdot 10^{-3}$
$\Sigma K_{zi} R_{zi}$	4.113314	$3.905 \cdot 10^{-3}$
$\Sigma K_{xi} R_{xi}$	4.112123	$3.902 \cdot 10^{-3}$
$K_{x3}$	1.000400	$1.591 \cdot 10^{-3}$
$K_{x4}$	1.001000	$2.345 \cdot 10^{-3}$
$K_{z3}$	0.999300	$1.591 \cdot 10^{-3}$
$K_{z4}$	1.000200	$2.345 \cdot 10^{-3}$
$M_{\text{Pb Assay1}}$	207.2103448 g/mol	$665.1 \cdot 10^{-6} \text{g/mol}$

**Uncertainty Budgets:**

**c<sub>x</sub>:** amount content of the sample x

Quantity	Value	Standard Uncertainty	Distribution	Sensitivity Coefficient	Uncertainty Contribution	Index
c <sub>z</sub>	0.09260484 μmol/g	27.77·10 <sup>-6</sup> μmol/g				
m <sub>y1</sub>	1.1360000 g	200.0·10 <sup>-6</sup> g	normal	0.047	9.5·10 <sup>-6</sup> μmol/g	0.3 %
m <sub>x</sub>	1.0440000 g	200.0·10 <sup>-6</sup> g	normal	-0.051	-10·10 <sup>-6</sup> μmol/g	0.3 %
m <sub>z</sub>	1.1029000 g	200.0·10 <sup>-6</sup> g	normal	0.049	9.7·10 <sup>-6</sup> μmol/g	0.3 %
m <sub>y2</sub>	1.0654000 g	200.0·10 <sup>-6</sup> g	normal	-0.050	-10·10 <sup>-6</sup> μmol/g	0.3 %
K <sub>y1</sub>	0.999900	1.335·10 <sup>-3</sup>				
R <sub>y1</sub>	640.00·10 <sup>-6</sup>	14.14·10 <sup>-6</sup>	normal	-0.077	-1.1·10 <sup>-6</sup> μmol/g	0.0 %
K <sub>b1</sub>	0.998700	1.335·10 <sup>-3</sup>				
R <sub>b1</sub>	0.2936000	258.1·10 <sup>-6</sup>	normal	0.21	55·10 <sup>-6</sup> μmol/g	9.3 %
K <sub>x1</sub>	0.999200	1.335·10 <sup>-3</sup>				
R <sub>x1</sub>	2.140200	1.909·10 <sup>-3</sup>	normal	-0.016	-31·10 <sup>-6</sup> μmol/g	2.9 %
K <sub>b2</sub>	0.998300	1.335·10 <sup>-3</sup>				
R <sub>b2</sub>	0.5050000	459.6·10 <sup>-6</sup>	normal	-0.14	-64·10 <sup>-6</sup> μmol/g	12.7 %
K <sub>z1</sub>	0.998900	1.335·10 <sup>-3</sup>				
R <sub>z1</sub>	2.142900	1.909·10 <sup>-3</sup>	normal	0.020	38·10 <sup>-6</sup> μmol/g	4.4 %
ΣK <sub>zi</sub> R <sub>zi</sub>	4.113314	3.905·10 <sup>-3</sup>				
ΣK <sub>xi</sub> R <sub>xi</sub>	4.112123	3.902·10 <sup>-3</sup>				
c <sub>blank</sub>	450.0·10 <sup>-9</sup> μmol/g	200.0·10 <sup>-9</sup> μmol/g	normal	-1.0	-200·10 <sup>-9</sup> μmol/g	0.0 %
K <sub>x2</sub>	1.0	0.0				
R <sub>x2</sub>	1.0					
K <sub>x3</sub>	1.000400	1.591·10 <sup>-3</sup>				
R <sub>x3</sub>	0.914200	1.131·10 <sup>-3</sup>	normal	0.013	15·10 <sup>-6</sup> μmol/g	0.7 %
K <sub>x4</sub>	1.001000	2.345·10 <sup>-3</sup>				
R <sub>x4</sub>	0.0590100	123.7·10 <sup>-6</sup>	normal	0.013	1.6·10 <sup>-6</sup> μmol/g	0.0 %
K <sub>z2</sub>	1.0	0.0				
R <sub>z2</sub>	1.0					
K <sub>z3</sub>	0.999300	1.591·10 <sup>-3</sup>				
R <sub>z3</sub>	0.914700	1.131·10 <sup>-3</sup>	normal	-0.013	-15·10 <sup>-6</sup> μmol/g	0.7 %
K <sub>z4</sub>	1.000200	2.345·10 <sup>-3</sup>				
R <sub>z4</sub>	0.0587000	123.7·10 <sup>-6</sup>	normal	-0.013	-1.6·10 <sup>-6</sup> μmol/g	0.0 %
M <sub>Pb Assay1</sub>	207.2103448 g/mol	665.1·10 <sup>-6</sup> g/mol				
M <sub>z1</sub>	207.976636000 g/mol	3.000·10 <sup>-6</sup> g/mol	normal	-130·10 <sup>-6</sup>	-400·10 <sup>-12</sup> μmol/g	0.0 %

Determination of the amount of lead in water using double isotope dilution and inductively coupled plasma mass spectrometry

Quantity	Value	Standard Uncertainty	Distribution	Sensitivity Coefficient	Uncertainty Contribution	Index
$M_{z2}$	205.974449000 g/mol	$3.000 \cdot 10^{-6}$ g/mol	normal	$-63 \cdot 10^{-6}$	$-190 \cdot 10^{-12}$ $\mu$ mol/g	0.0 %
$M_{z3}$	206.975880000 g/mol	$3.000 \cdot 10^{-6}$ g/mol	normal	$-58 \cdot 10^{-6}$	$-170 \cdot 10^{-12}$ $\mu$ mol/g	0.0 %
$M_{z4}$	203.973028000 g/mol	$3.000 \cdot 10^{-6}$ g/mol	normal	$-3.7 \cdot 10^{-6}$	$-11 \cdot 10^{-12}$ $\mu$ mol/g	0.0 %
$m_2$	1.0292000 g	$200.0 \cdot 10^{-6}$ g	normal	0.052	$10 \cdot 10^{-6}$ $\mu$ mol/g	0.3 %
$d_2$	99.93100 g	0.01000 g	normal	$-540 \cdot 10^{-6}$	$-5.4 \cdot 10^{-6}$ $\mu$ mol/g	0.0 %
$m_1$	0.36544000 g	$50.00 \cdot 10^{-6}$ g	normal	0.15	$7.4 \cdot 10^{-6}$ $\mu$ mol/g	0.2 %
w	0.999990000 g/g	$5.000 \cdot 10^{-6}$ g/g	normal	0.054	$270 \cdot 10^{-9}$ $\mu$ mol/g	0.0 %
$d_1$	196.14000 g	0.03000 g	normal	$-270 \cdot 10^{-6}$	$-8.2 \cdot 10^{-6}$ $\mu$ mol/g	0.2 %
$k_{mol}$	$1.0 \cdot 10^6$ $\mu$ mol/mol					
$K_{0\_b1}$	0.9987000	$883.9 \cdot 10^{-6}$	normal	0.062	$55 \cdot 10^{-6}$ $\mu$ mol/g	9.4 %
$K_{bias\_b1}$	0.0	$1.000 \cdot 10^{-3}$	normal	0.062	$62 \cdot 10^{-6}$ $\mu$ mol/g	12.1 %
$K_{0\_b2}$	0.9983000	$883.9 \cdot 10^{-6}$	normal	-0.070	$-62 \cdot 10^{-6}$ $\mu$ mol/g	12.0 %
$K_{bias\_b2}$	0.0	$1.000 \cdot 10^{-3}$	normal	-0.070	$-70 \cdot 10^{-6}$ $\mu$ mol/g	15.3 %
$K_{0\_x1}$	0.9992000	$883.9 \cdot 10^{-6}$	normal	-0.034	$-30 \cdot 10^{-6}$ $\mu$ mol/g	2.8 %
$K_{bias\_x1}$	0.0	$1.000 \cdot 10^{-3}$	normal	-0.034	$-34 \cdot 10^{-6}$ $\mu$ mol/g	3.6 %
$K_{0\_x2}$	1.0					
$K_{bias\_x2}$	0.0					
$K_{0\_x3}$	1.000400	$1.237 \cdot 10^{-3}$	normal	0.012	$15 \cdot 10^{-6}$ $\mu$ mol/g	0.7 %
$K_{bias\_x3}$	0.0	$1.000 \cdot 10^{-3}$	normal	0.012	$12 \cdot 10^{-6}$ $\mu$ mol/g	0.4 %
$K_{0\_x4}$	1.001000	$2.121 \cdot 10^{-3}$	normal	$770 \cdot 10^{-6}$	$1.6 \cdot 10^{-6}$ $\mu$ mol/g	0.0 %
$K_{bias\_x4}$	0.0	$1.000 \cdot 10^{-3}$	normal	$770 \cdot 10^{-6}$	$770 \cdot 10^{-9}$ $\mu$ mol/g	0.0 %
$K_{0\_y1}$	0.9999000	$883.9 \cdot 10^{-6}$	normal	$-49 \cdot 10^{-6}$	$-44 \cdot 10^{-9}$ $\mu$ mol/g	0.0 %
$K_{bias\_y1}$	0.0	$1.000 \cdot 10^{-3}$	normal	$-49 \cdot 10^{-6}$	$-49 \cdot 10^{-9}$ $\mu$ mol/g	0.0 %
$K_{0\_z1}$	0.9989000	$883.9 \cdot 10^{-6}$	normal	0.042	$37 \cdot 10^{-6}$ $\mu$ mol/g	4.3 %
$K_{bias\_z1}$	0.0	$1.000 \cdot 10^{-3}$	normal	0.042	$42 \cdot 10^{-6}$ $\mu$ mol/g	5.5 %
$K_{0\_z2}$	1.0					
$K_{bias\_z2}$	0.0					
$K_{0\_z3}$	0.999300	$1.237 \cdot 10^{-3}$	normal	-0.012	$-15 \cdot 10^{-6}$ $\mu$ mol/g	0.7 %
$K_{bias\_z3}$	0.0	$1.000 \cdot 10^{-3}$	normal	-0.012	$-12 \cdot 10^{-6}$ $\mu$ mol/g	0.4 %
$K_{0\_z4}$	1.000200	$2.121 \cdot 10^{-3}$	normal	$-750 \cdot 10^{-6}$	$-1.6 \cdot 10^{-6}$ $\mu$ mol/g	0.0 %
$K_{bias\_z4}$	0.0	$1.000 \cdot 10^{-3}$	normal	$-750 \cdot 10^{-6}$	$-750 \cdot 10^{-9}$ $\mu$ mol/g	0.0 %
$c_x$	0.0537374 $\mu$ mol/g	$179.9 \cdot 10^{-6}$ $\mu$ mol/g				

**Results:**

Quantity	Value	Expanded Uncertainty	Coverage factor	Coverage
$c_x$	0.05374 $\mu\text{mol/g}$	$180 \cdot 10^{-6} \mu\text{mol/g}$	1.00	manual